

IN THE CLAIMS

Please amend claims 4, 5, 10, 11, 12, and 13.

1. (Previously Canceled)
2. (Previously Canceled)
3. (Previously Canceled)
4. (Currently Amended) A process for forming a hydrogen storage material comprising:
 - providing a sodium aluminum hydride;
 - mixing [a] about 1%/wt to about 4%/wt of titanium to said sodium aluminum hydride; and,
 - supplying a combination of heat and pressure in the presence of hydrogen gas sufficient to melt said sodium aluminum hydride and titanium mixture, thereby providing a fused hydrogen storage material having a hydrogen release point at normal atmospheres of between about 50°C to about 90°C.
5. (Currently Amended) A process of forming a hydrogen storage material comprising:
 - supplying at least one complex hydride;
 - mixing with said complex hydride a dopant selected from the group consisting of ~~metals comprising~~ titanium, zirconium, vanadium, iron, cobalt, nickel, lanthanum, and mixtures thereof;
 - subjecting said mixture of complex hydride and said dopant under pressure in the presence of hydrogen gas;
 - raising the temperature of said mixture of said complex hydride and said dopant and said hydrogen gas to a melting point of said complex hydride; and,
 - maintaining said heat and pressure for a time sufficient to form a fused product, wherein said fused product has a reversible ability to store and release hydrogen.

6. (Previously Added) The process according to claim 5 wherein said at least one complex hydride is sodium aluminum hydride.
7. (Previously Added) The process according to claim 5 wherein said at least one complexhydride comprises lithium hydride.
8. (Previously Added) The process according to claim 5 wherein said at least one complexhydride comprises sodium hydride.
9. (Previously Added) The process according to claim 5 wherein said at least one complex hydride comprises a mixture of sodium aluminum hydride, lithium hydride, and sodium hydride.
10. (Currently Amended) The process according to claim 9 wherein said sodium aluminum hydride, said lithium hydride, and said sodium hydride are present in ~~substantially~~ about equi-molar amounts.
11. (Currently Amended) The ~~product~~ hydrogen storage material of the process according to claim 4.
12. (Currently Amended) The ~~product~~ hydrogen storage material of the process according to claim 5.
13. (Currently Amended) The ~~product~~ hydrogen storage material of the process according to claim 9.